

# Selectively Substituted Thiophenes and Indoles by a Tandem Palladium-Catalyzed Multicomponent Reaction

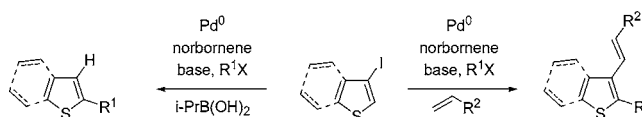
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## ABSTRACT



A variety of di- and trisubstituted thiophenes were synthesized by a one-pot palladium-catalyzed ortho-alkylation sequence terminated by either Heck or C–H coupling. Initial results toward the functionalization of indoles are also presented.

Polysubstituted thiophenes are useful compounds for both pharmaceutical<sup>1</sup> and material sciences.<sup>2</sup> While a number of methods have been reported for their preparation, selective substitution remains as a challenging goal.<sup>1,5</sup> Various sequential substitution reactions have been reported for aromatic compounds, most notably using palladium catalysis;<sup>3,4</sup> however, there are few reports in this area pertaining to

thiophenes.<sup>5</sup> We are interested in the sequential substitution of aromatic compounds and recently reported a method for the formation of fused aromatic rings<sup>6</sup> based on the work of Catellani and co-workers.<sup>7</sup>

We have extended the ortho-alkylation procedure to include heteroaryl halides and now report a route to selectively functionalized thiophenes.

Initial studies found that 2-iodothiophene is a poor substrate for the tandem ortho-alkylation/Heck coupling reaction. Thus, we examined commercially available 3-iodothiophene (**1**) as a substrate (Table 1). Among several parameters tested, Pd(OAc)<sub>2</sub> and tri-2-furylphosphine (TFP) in CH<sub>3</sub>CN gave the best yield of the desired product **2a** compared to the direct Heck product **3**.<sup>8,9</sup> The amount of norbornene greatly affects the yield of **2a**, improving with

(1) (a) Press, J. B. In *The Chemistry of Heterocyclic Compounds: Thiophene and Its Derivatives*; Gronowitz, S., Ed.; John Wiley & Sons: New York, 1991; Vol. 44, Part 4, pp 397–502. (b) Pillai, A. D.; Rathod, P. D.; Xavier, F. P.; Vasu, K. K.; Padh, H.; Sudarsanam, V. *Bioorg. Med. Chem.* **2004**, *12*, 4667–4671. (c) Sall, D. J.; Bastian, J. A.; Briggs, S. L.; Buben, J. A.; Chirgadze, N. Y.; Clawson, D. K.; Denney, M. L.; Giera, D. D.; Gifford-Moore, D. S.; Harper, R. W.; Hauser, K. L.; Klimkowski, V. J.; Kohn, T. J.; Lin, H.-S.; McCowan, J. R.; Palkowitz, A. D.; Smith, G. F.; Takeuchi, K.; Thrasher, K. J.; Tinsley, J. M.; Utterback, B. G.; Yan, S.-C. B.; Zhang, M. *J. Med. Chem.* **1997**, *40*, 3489–3493.

(2) (a) Inganaes, O.; Berggren, M.; Andersson, M. R.; Gustafsson, G.; Hjertberg, T.; Wennerstroem, O.; Dyreklev, P.; Granstroem, M. *Synth. Met.* **1995**, *71*, 2121–2124. (b) Wolf, M. O. *Adv. Mater.* **2001**, *13*, 545–553.

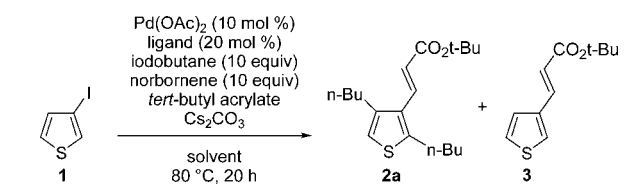
(3) (a) Hegedus, L. S. In *Transition Metals in the Synthesis of Complex Organic Molecules*, 2nd ed.; University Science Books: Sausalito, CA, 1999. (b) Tsuji, J. *Palladium Reagents and Catalysts*; John Wiley: Chichester, 2004.

(4) (a) Poli, G.; Giambastiani, G.; Heumann, A. *Tetrahedron* **2000**, *56*, 5959–5989. (b) Negishi, E.; Copéret, C.; Ma, S.; Liou, S.-Y.; Liu, F. *Chem. Rev.* **1996**, *96*, 365–394.

(5) Hassan, J.; Gozzi, C.; Schulz, E.; Lemaire, M. *J. Organomet. Chem.* **2003**, *687*, 280–283 and references therein.

(6) (a) Lautens, M.; Piguel, S. *Angew. Chem., Int. Ed.* **2000**, *39*, 1045–1046. (b) Lautens, M.; Paquin, J.-F.; Piguel, S.; Dahlmann, M. *J. Org. Chem.* **2001**, *66*, 8127–8134. (c) Lautens, M.; Paquin, J.-F.; Piguel, S. *J. Org. Chem.* **2002**, *67*, 3972–3974. (d) Pache, S.; Lautens, M. *Org. Lett.* **2003**, *5*, 4827–4830. (e) Alberico, D.; Paquin, J.-F.; Lautens, M. *Tetrahedron* **2005**, *61*, 6283–6297. (f) Bressy, C.; Alberico, D.; Lautens, M. *J. Am. Chem. Soc.* **2005**, *127*, 13148–13149.

(7) (a) Catellani, M.; Fagnola, M. C. *Angew. Chem., Int. Ed. Engl.* **1994**, *33*, 2421–2422. (b) Catellani, M.; Frignani, F.; Rangoni, A. *Angew. Chem., Int. Ed. Engl.* **1997**, *36*, 119–122. (c) Faccini, F.; Motti, E.; Catellani, M. *J. Am. Chem. Soc.* **2004**, *126*, 78–79.

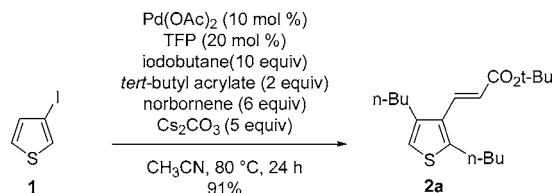
**Table 1.** Optimization of Domino Alkylation/Heck Reaction

entry	solvent	ligand	<i>tert</i> -butyl acrylate (equiv)	Cs <sub>2</sub> CO <sub>3</sub> (equiv)	yield <sup>a</sup> (%)	
					<b>2a</b>	<b>3</b>
1 <sup>b</sup>	DME	PPh <sub>3</sub>	5	2	<1	46
2 <sup>b</sup>	DME	TFP	5	2	<1	43
3 <sup>b</sup>	Dioxane	TFP	5	2	<1	59
4 <sup>b</sup>	CH <sub>3</sub> CN	TFP	5	2	<1	56
5	CH <sub>3</sub> CN	TFP	5	2	37	26
6	DME	PPh <sub>3</sub>	2	5	75	0
7	DME	TFP	2	5	69	0
8	CH <sub>3</sub> CN	TFP	2	5	80	0
9	CH <sub>3</sub> CN	PPh <sub>3</sub>	2	5	56	0

<sup>a</sup> Isolated yields. <sup>b</sup> With 2 equiv of norbornene.

increasing equivalents of norbornene and reaching a maximum at 6 equivalents.

The optimal conditions of **1** (1.0 equiv), 1-iodobutane (10 equiv), and *tert*-butyl acrylate (2.0 equiv) at 80 °C in CH<sub>3</sub>CN in the presence of Pd(OAc)<sub>2</sub> (10 mol %), TFP (20 mol %), norbornene (6.0 equiv), and Cs<sub>2</sub>CO<sub>3</sub> (5.0 equiv) gave the trisubstituted product **2a** in 91% yield (Scheme 1). The use of 3-bromothiophene under the same reaction conditions gave **2a** in 33% yield.

**Scheme 1.** Optimal Conditions for the Domino Ortho-Alkylation/Heck Reaction

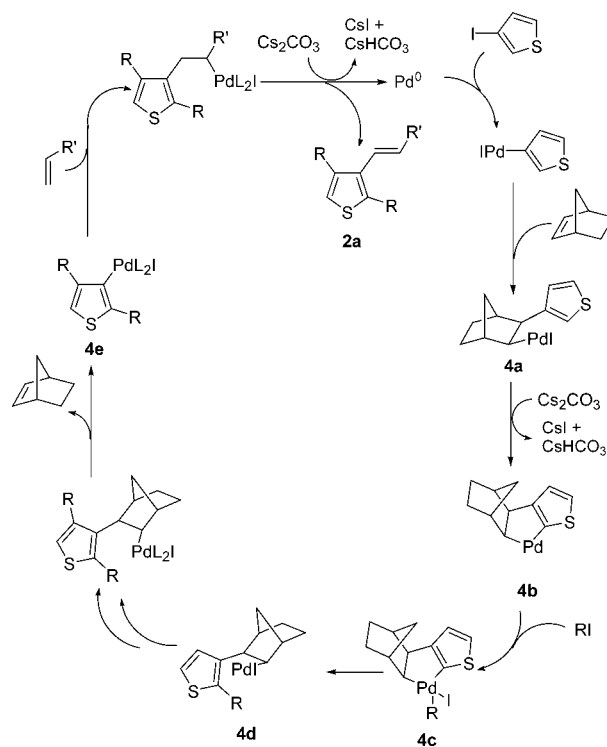
The proposed mechanism is presented in Scheme 2.<sup>7,10</sup>

Pd(0) inserts into the C–I bond of 3-iodothiophene, followed by carbopalladation of norbornene to give **4a**. C–H activation of the 2-position forms **4b**. Oxidative addition of the alkyl iodide leads to the Pd(IV) species **4c** and reductive elimination generates **4d**. This process then occurs for the 4-position C–H bond, and extrusion of norbornene gives the 2,4-dialkylated 3-(thienyl)palladium(II) complex **4e**.

(8) The use of Pd(OAc)<sub>2</sub> as a palladium source gave the best selectivity. The reaction using Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> or PdCl<sub>2</sub>(CH<sub>3</sub>CN)<sub>2</sub> as a catalyst precursor gave the direct Heck product as the major product together with trace amounts of the desired compound.

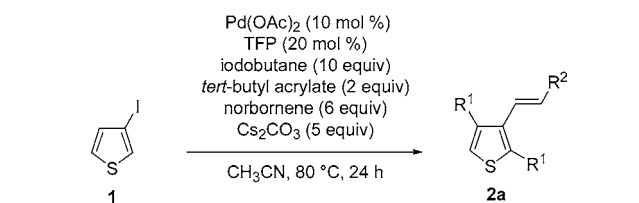
(9) The use of PPh<sub>3</sub> also gives **2a** in comparable but lower yields.

(10) This mechanism is based on mechanistic studies of Catellani, Pregosin, and co-workers (ref 6c,d) and on our own preliminary studies.

**Scheme 2.** Proposed Mechanism

Alternatively, C–H activation of the 4-position could take place in the first step; however, at this time we do not know which C–H activation occurs first. Heck reaction with **4e** yields product **2a**. We believe an excess of norbornene is needed to compete with the direct Heck pathway.

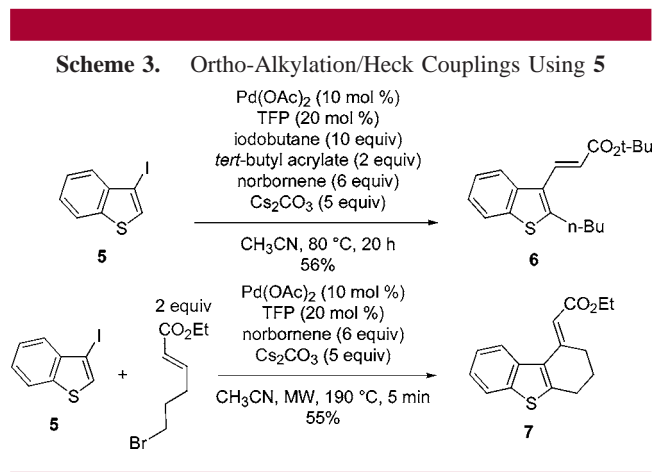
The scope of this reaction is presented in Table 2.<sup>11</sup> In all cases, trisubstituted products were obtained in moderate to high yields. In addition to *tert*-butyl acrylate, ethyl and

**Table 2.** Scope of the Ortho-Alkylation/Heck Coupling

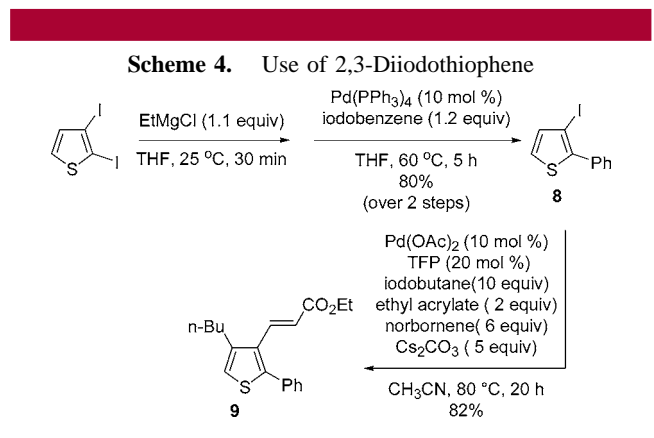
entry	R <sup>1</sup>	R <sup>2</sup>	<b>2</b> (yield, %) <sup>a</sup>
1	<i>n</i> -Bu	CO <sub>2</sub> - <i>t</i> -Bu	<b>2a</b> (91)
2	<i>n</i> -Bu	CO <sub>2</sub> Et	<b>2b</b> (96)
3	<i>n</i> -Bu	CO <sub>2</sub> Me	<b>2c</b> (65) <sup>b</sup>
4	<i>n</i> -Bu	C(O)NMe <sub>2</sub>	<b>2d</b> (70) <sup>c</sup>
5	<i>n</i> -Bu	C(O)NH- <i>t</i> -Bu	<b>2e</b> (54) <sup>b</sup>
6	<i>n</i> -Bu	C(O)Et	<b>2f</b> (52)
7	TBSO(CH <sub>2</sub> ) <sub>3</sub>	CO <sub>2</sub> Et	<b>2g</b> (74)
8	Ph(CH <sub>2</sub> ) <sub>3</sub>	CO <sub>2</sub> Et	<b>2h</b> (82) <sup>b</sup>
9	Cl(CH <sub>2</sub> ) <sub>3</sub>	CO <sub>2</sub> - <i>t</i> -Bu	<b>2i</b> (55) <sup>c</sup>

<sup>a</sup> Isolated yields. <sup>b</sup> With 4 equiv of norbornene. <sup>c</sup> With 10 equiv of norbornene.

methyl acrylate can be used as Heck acceptors (entries 1–3), as well as acrylamides (entries 4–5) and ethylvinyl ketone (entry 6). A variety of iodides with remote reactive functional groups were also tolerated under the reaction conditions (entries 7–9). The ortho-alkylation/Heck coupling reaction was further extended to 3-iodobenzo[*b*]thiophene (**5**) to give the adduct **6** in 56% yield. Using a bis-acceptor produced tricyclic compound **7** in 55% yield (Scheme 3). To the best of our knowledge, these are the first examples of Heck reactions using 3-iodobenzo[*b*]thiophenes.



To create an unsymmetrically substituted thiophene at the 2- and 4-positions, regioselective halogen–magnesium exchange<sup>12</sup> of 2,3-diiodothiophene with EtMgCl (1.1 equiv) produced the 2-substituted Grignard reagent,<sup>13</sup> which underwent Kumada–Tamao–Corriu coupling<sup>14</sup> with iodobenzene (1.2 equiv) and Pd(PPh<sub>3</sub>)<sub>4</sub> (10 mol %) to give **8** in 80% yield (Scheme 4). Subjecting **8** to our optimized conditions gave unsymmetrically substituted thiophene **9** in 82% yield.



To generate 2,4-disubstituted thiophenes, an ortho-alkylation/C–H coupling previously developed in our group for iodoarenes<sup>15</sup> was applied to 3-iodothiophene to produce **10** (Table 3). Compared to arenes, lower temperatures and longer reaction times were needed to inhibit thiophene polymerization and decomposition. The mechanism for this transformation is similar to Scheme 2, although the terminat-

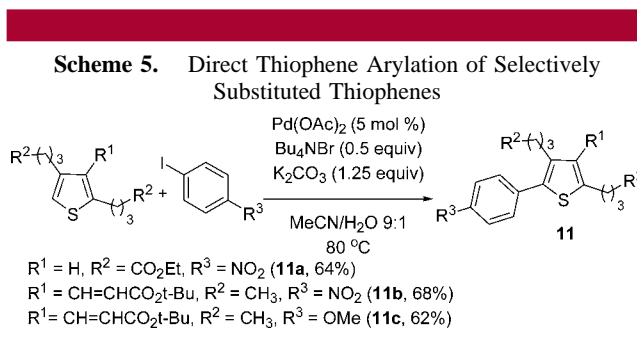
**Table 3.** Scope of Ortho-Alkylation/C–H Coupling

entry	R	yield <sup>a</sup> (%)
1	CO <sub>2</sub> Et	<b>10a</b> (42)
2	1-oxirane	<b>10b</b> (54)
3	2-(1,3)-dioxolane	<b>10c</b> (36)
4	<i>N</i> -phthalimide	<b>10d</b> (24)
5	CN	<b>10e</b> (26)

<sup>a</sup> Isolated yields. <sup>b</sup> Slow addition.

ing step from **4e** is a C–H coupling using isopropylboronic acid. Though the yields are moderate to low, a variety of functional groups are tolerated, and afford products which are difficult to synthesize using classical methods. The functionalized alkyl chains can also be useful synthetic handles for further modification.

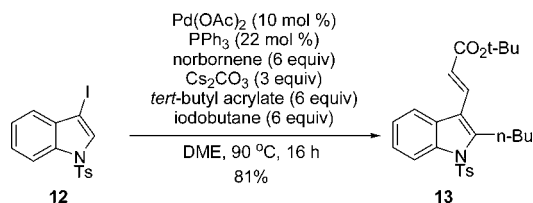
To create tri- and tetrasubstituted thiophenes, the 5-position was functionalized using direct thiophene arylation<sup>16</sup> to generate a 2,4,5-trisubstituted thiophene (**11a**) and tetra-substituted thiophenes (**11b** and **11c**) (Scheme 5) in good yields from the parent compounds.



Our next goal was to extend the methodology to the functionalization of indoles, a structural motif that is very common in a number of natural products and medically important compounds. After careful screening of the reaction parameters, we were able to achieve 81% yield of the desired

(11) Typical procedure for sequential alkylation/alkenylation reaction of 3-iodothiophene (Table 2, entry 1): A mixture of 3-iodothiophene (42 mg, 0.2 mmol), 1-iodobutane (368 mg, 2.0 mmol), *tert*-butyl acrylate (53 mg, 0.4 mmol), Pd(OAc)<sub>2</sub> (4.5 mg, 10 mol %), TFP (9.3 mg, 20 mol %), norbornene (113 mg, 1.2 mmol), and Cs<sub>2</sub>CO<sub>3</sub> (326 mg, 1.0 mmol) in dry CH<sub>3</sub>CN (2.0 mL) was stirred at 80 °C for 20 h under nitrogen. After being cooled to room temperature, the resulting mixture was treated with water and extracted with Et<sub>2</sub>O. The combined organic phase was dried over MgSO<sub>4</sub> and concentrated under reduced pressure. The residue was chromatographed on silica gel (hexane/AcOEt = 20/1–10/1) to afford **2a** (59 mg, 91%) as a pale yellow oil.

**Scheme 6.** Optimized Coupling for 2,3 Coupling of 3-Iodoindole



product **13** (Scheme 6). Fortuitously, it could be crystallized which provided a method to unequivocally prove the structure of **13**. During the course of optimization, it was found that the nature of the nitrogen protecting group was crucial to the efficiency of the reaction. Use of a methyl protecting group yielded only the direct Heck product.

In summary, we have established a palladium-catalyzed tandem alkylation/alkenylation reaction of 3-iodothiophene to construct trisubstituted thiophenes in good yields. In

(12) Knochel and co-workers reported that polyfunctional arylmagnesium and alkenylmagnesium reagents can be prepared by iodine–magnesium exchange reaction: Knochel, P.; Dohle, W.; Gommernann, N.; Kneisel, F. F.; Kopp, F.; Korn, T.; Sapountzis, I.; Vu, V. A. *Angew. Chem., Int. Ed.* **2003**, *36*, 4302–4320 and references therein.

(13) Regioselective halogen–magnesium exchange reaction of 2,3-dibromothiophene was reported; see: Christophersen, C.; Begtrup, M.; Ebdrup, S.; Petersen, H.; Vedsø, P. *J. Org. Chem.* **2003**, *68*, 9513–9516.

(14) For reviews on Kumada–Tamao coupling, see: (a) Tamao, K. *J. Organomet. Chem.* **2002**, *653*, 27–33. (b) Hayashi, T. In *Handbook of Organopalladium Chemistry for Organic Synthesis*; Negishi, E.-I., Ed.; John & Wiley & Sons: Hoboken, NJ; 2002; Vol. 1, pp 791–806.

(15) Wilhelm, T.; Lautens, M. *Org. Lett.* **2005**, *7*, 4053–4056.

addition, 3-iodobenzo[*b*]thiophene and 2-substituted 3-iodothiophene can be used. Application of our ortho-alkylation/C–H coupling produces 2,4-disubstituted thiophenes, and direct arylation of di- and trisubstituted thiophenes produces 2,4,5-trisubstituted and tetrasubstituted thiophene derivatives. Finally, this methodology has also been extended to include indoles, the scope of which is currently under investigation and will be reported in due course.

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**Supporting Information Available:** Experimental procedures and characterization data for new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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(16) (a) Gozzi, C.; Lavenot, L.; Ilg, K.; Penalva, V.; Lemaire, M. *Tetrahedron Lett.* **1997**, *51*, 8867–8870. (b) Okazawa, T.; Satoh, T.; Miura, M.; Nomura, M. *J. Am. Chem. Soc.* **2002**, *124*, 5286–5287. (c) Lavenot, L.; Gozzi, C.; Ilg, K.; Orlova, I.; Penalva, V.; Lemaire, M. *J. Organomet. Chem.* **1998**, *567*, 49–55. (d) Glover, B.; Harvey, K. A.; Liu, B.; Sharp, M. J.; Tymoschenko, M. F. *Org. Lett.* **2003**, *5*, 301–304.